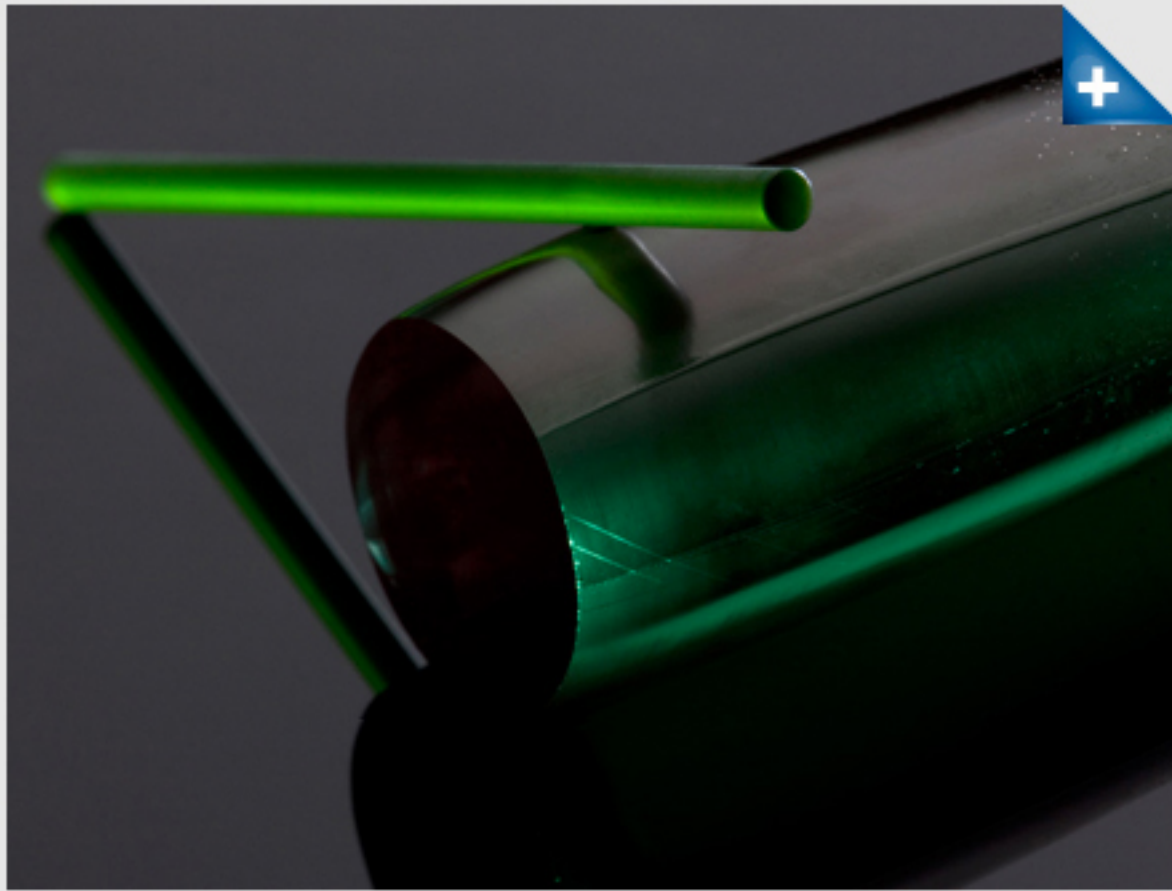


## Laser Materials CTH:YAG



### General Information

Triple doped  $\text{Cr}^{3+}$ ,  $\text{Tm}^{3+}$ ,  $\text{Ho}^{3+}$ :YAG is an efficient solid-state laser medium for 2097nm generation, widely used in military, medicine, and remote sensing applications. High spectral overlap of pump radiation (lamp or diode) with the  $\text{Cr}^{3+}$  and  $\text{Tm}^{3+}$  absorption bands, and a highly efficient conversion from the absorption bands into the  $^5\text{I}_7 \rightarrow ^5\text{I}_8$   $\text{Ho}^{3+}$  emission band, enables 2 micron laser architectures with high quantum efficiency. [1], [2]

SMC provides high quality CTH:YAG for your prouction and/or development efforts.

[Contact us](#) with your specific requirements or for availability and pricing.

### Dopant Ion

Cr3+ Concentration	0.85%
Tm3+ Concentration	5.9%
Ho3+ Concentration	0.36%

### Common Operating Specs

Emission Wavelength	2.080 $\mu\text{m}$
Laser Transition	$^5\text{I}_7 \rightarrow ^5\text{I}_8$
Flourescence Lifetime	8.5 ms
Pump Wavelength	flash lamp or diode pumped @ 780nm

### Physical Properties

Coefficient of Thermal Expansion	$6.14 \times 10^{-6} \text{ K}^{-1}$
Thermal Diffusivity	$0.041 \text{ cm}^2 \text{ s}^{-2}$
Thermal Conductivity	$11.2 \text{ W m}^{-1} \text{ K}^{-1}$
Specific Heat (Cp)	$0.59 \text{ J g}^{-1} \text{ K}^{-1}$
Thermal Shock Resistant	$800 \text{ W m}^{-1}$
Refractive Index @ 632.8 nm	1.83
dn/dT (Thermal Coefficient of Refractive Index) @ 1064nm	$7.8 \times 10^{-6} \text{ K}^{-1}$
Molecular Weight	$593.7 \text{ g mol}^{-1}$
Melting Point	1965°C
Density	$4.56 \text{ g cm}^{-3}$
MOHS Hardness	8.25
Young's Modulus	335 Gpa
Tensile Strength	2 Gpa
Crystal Structure	Cubic
Standard Orientation	<111>
Y3+ Site Symmetry	$\text{D}_2$
Lattice Constant	$a=12.013 \text{ \AA}$

### References

1) CTH:YAG was initially suggested by, B. M. Antipenko, et. al. , Zh. Tekh. Fiz. Pis'ma, 11, 682 (1985)

2) An overview of the cross relaxation energetics is given by, A. A. Kaminskii, "Crystalline Lasers: Physcial Processes and Operating Schemes", CRC Press, New York, (1996), ISBN 0-8493-3720-8. Chapter 8.2.

### Absorption Coefficient Chart

